Exploring compositional and mesostructural effects of Mn-doped Co_3O_4 spinel catalysts on the catalytic 2-propanol oxidation

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The influence of Mn incorporation into three $Co_{3-x}Mn_xO_4$ series with unique morphologies and its influence on the electronic structure and catalytic activity was investigated. Two catalyst synthesis approaches were employed namely the crystalline precursor decomposition approach^[1] and the hard-templating method^[2]. Single-phase $Co_{3-x}Mn_xO_4$ catalysts with platelet morphologies (Fig. 1; left) were synthesized via pH- and temperature-controlled co-precipitation of hydroxide precursors and subsequent thermal decomposition. Catalysts with spherical morphologies (Fig. 1; middle) were prepared by applying the same approach using hydroxycarbonate precursors. In addition, phase-pure $Co_{3-x}Mn_xO_4$ nanowires (Fig. 1; right) were prepared by nanocasting route using SBA-15 silica as a template. When performing nitrogen physisorption, all three sample series showed increasing surface areas with increasing Mn content, with the nanowires achieving the highest surface areas of 130 m²/g among the series.



Figure 1: SEM images of the different $Co_{3-x}Mn_xO_4$ spinel catalysts. Platelets (left), spherical particles (middle) and nanowires (right).

The catalytic activity and selectivity of these catalysts were then measured in gasphase 2-propanol oxidation. Preliminary results have already shown promising catalytic activity trends in the dependency of Mn content and morphology.

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